BIOLOGY 2 BIOCHEMISTRY ACTIVITY #6

NAME			
NAMF	HOUR		

PROTEINS

GENERAL CHARACTERISTICS AND IMPORTANCES:		
BUILDING BLOCKS:		
	,	
ANION	CATION	DIPOLAR ION

CLASSIFICATION:	
Nonpolar: H ₃ N ⁺ - C - C O CH ₂ CH ₃ CH ₃	POLAR: H ₃ N ⁺ - C - C O - C O O - C O O O O O O O O O
POLAR CHARGED ACIDIC: H ₃ N ⁺ - C - C C C C C C C C C C C C C C C C	POLAR CHARGED BASIC: H H CH CH CH CH CH CH CH CH
PEPTIDE BONDS:	

PROTEIN CONFORMATION:	
PRIMARY:	
Val His Leu Lys Tyr His	
SECONDARY:	
99	
9999	
77.397	
8 4480	
0	
TERTIARY:	
(B)	
(15605)	
256	
(50°)	
Quaternary:	
QUATERNARY.	

DENATURATION:			

QUESTIONS:

1. Classify each of the following amino acids as **nonpolar**, **polar uncharged**, **polar charged acidic**, or **polar charged basic**.

a._____

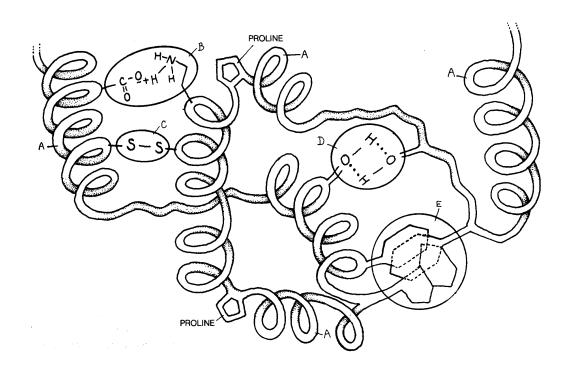
b.

C.

d._____

2. Draw a peptide bond between two amino acids.

3. Use the drawing below to answer the questions that follow.



- a. What level of protein structure is shown in the picture? _____
- b. Match the following with the correct letter from the diagram.

 $\underline{}$ α helix

_____ Disulfide bridge

_____ Hydrogen bonding

_____ Hydrophobic interaction

_____ Ionic bond

Indicate the level the following.	of protection structure (1, 2, 3, or 4) described in each of
α helix	
β pleated	sheets
Collagen	and hemoglobin
Determin	ed by the sequence of DNA bases
Form sta	bilized by hydrogen bonds
	bilized by hydrogen bonds, ionic bonds, hydrophobic actions, and disulfide bridges proteins
Interaction	on among several polypeptide chains
Most enz	ymes
Regular,	repeated folding of the peptide chain
Sequence	e of amino acids in a protein
What happens to	a protein when it is denatured?
How does denatu	ration affect the function of a protein? Why?
Explain how each	of the following causes a protein to denature.
Subjecting the protein to high temperature	
Placing the protein in a strong acid	
Placing the protein in an organic solvent	